

# CHEMISTRY :

# MEASUREMENT PRACTICE III

**A. Determine the number of significant digits in the following numbers:**

- \_\_\_\_\_ 1) 5600  
\_\_\_\_\_ 2) 45.00  
\_\_\_\_\_ 3) 105.0  
\_\_\_\_\_ 4) 0.00565  
\_\_\_\_\_ 5) 0.005400  
\_\_\_\_\_ 6) 89.543  
\_\_\_\_\_ 7) 5, 056, 300  
\_\_\_\_\_ 8) 95.0540  
\_\_\_\_\_ 9) 93,000,000

**B. Perform the indicated operations and express your answer to the correct number of significant digits:**

- \_\_\_\_\_ 10)  $(6.92)(7.9)$   
\_\_\_\_\_ 11)  $(8.245)(9.00)$   
\_\_\_\_\_ 12)  $(4.46)/(6.52)$   
\_\_\_\_\_ 13)  $(9.825)/(8.20)$   
\_\_\_\_\_ 14)  $(8.95) (9.162)/(4.25) (6.3)$

**C. Perform the indicated operations and express your answer to the correct number of significant digits:**

- \_\_\_\_\_ 15)  $5.50 + 0.528 + 9.2$   
\_\_\_\_\_ 16)  $420 + 8900 + 620$   
\_\_\_\_\_ 17)  $0.00526 - 0.52$   
\_\_\_\_\_ 18)  $820.0 + 19.5 + 6$   
\_\_\_\_\_ 19)  $4,285.75 - 520.1 - 386.255$   
\_\_\_\_\_ 20)  $(0.526) (895) + 20.8$   
\_\_\_\_\_ 21)  $3.414 \text{ s} + 10.02 \text{ s} + 58.325 \text{ s} + 0.000 \text{ 98 s}$   
\_\_\_\_\_ 22)  $1884 \text{ kg} + 0.94 \text{ kg} + 1.0 \text{ kg} + 9.778 \text{ kg}$   
\_\_\_\_\_ 23)  $2104.1 \text{ m} - 463.09 \text{ m}$   
\_\_\_\_\_ 24)  $2.326 \text{ h} - 0.104 \text{ 08 h}$   
\_\_\_\_\_ 25)  $10.19 \text{ m} \times 0.013 \text{ m}$

- \_\_\_\_\_ 26)  $140.01 \text{ cm} \times 26.042 \text{ cm} \times 0.0159 \text{ cm}$   
\_\_\_\_\_ 27)  $80.23 \text{ m} \div 2.4 \text{ s}$   
\_\_\_\_\_ 28)  $4.301 \text{ kg} \div 1.9 \text{ cm}^3$   
\_\_\_\_\_ 29)  $3.68 + 7.3645 + 0.5$   
\_\_\_\_\_ 30)  $0.243 + 76.720 + 4.6494$   
\_\_\_\_\_ 31)  $14.745 - 1.60$   
\_\_\_\_\_ 32)  $0.5642 - 0.260$   
\_\_\_\_\_ 33)  $6.02 \times 2.0$   
\_\_\_\_\_ 34)  $0.65 \times 427$   
\_\_\_\_\_ 35)  $0.022 \times 0.467$   
\_\_\_\_\_ 36)  $174 \div 24$   
\_\_\_\_\_ 37)  $420 \div 17.5$   
\_\_\_\_\_ 38)  $3.0899 \text{ mm} \times 22.4 \text{ mm}$   
\_\_\_\_\_ 39)  $3.4500 \text{ cm}^2 \div 450 \text{ cm}$   
\_\_\_\_\_ 40)  $13.80 \text{ cm} - 6.0741 \text{ cm}$

**D. For each item below determine the number of significant digits in the number or answer to the problem: (don't do the calculation)**

- \_\_\_\_\_ 41) 804.58  
\_\_\_\_\_ 42) 250.00  
\_\_\_\_\_ 43) 3000  
\_\_\_\_\_ 44)  $10.00 \text{ m} \times 84.767 \text{ m}$   
\_\_\_\_\_ 45) 0.00300900870  
\_\_\_\_\_ 46)  $400 \times 87,488$   
\_\_\_\_\_ 47) 180.0001  
\_\_\_\_\_ 48)  $3.0 \times 4.000$   
\_\_\_\_\_ 49) 0.00560  
\_\_\_\_\_ 50)  $0.7600 \div 1.50$

# CHEMISTRY

# Scientific Notation Practice II

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A. Express the following numbers in correct scientific notation:

\_\_\_\_\_ 1. 8 720 000

\_\_\_\_\_ 2. 0.00 000 628

B. Rewrite the following numbers so they are in correct scientific notation:

\_\_\_\_\_ 3.  $652.9 \times 10^5$

\_\_\_\_\_ 4.  $0.00\ 598 \times 10^{-15}$

\_\_\_\_\_ 5.  $0.0\ 259 \times 10^8$

C. Carry out the indicated operations and express your answer in correct scientific notation:

\_\_\_\_\_ 6.  $(3.04 \times 10^{22}) + (3.04 \times 10^{23})$

\_\_\_\_\_ 7.  $(6.54 \times 10^5) - (2.0 \times 10^3)$

\_\_\_\_\_ 8.  $(2.5 \times 10^3) (3.62 \times 10^5)$

\_\_\_\_\_ 9.  $(9.12 \times 10^1) (6.55 \times 10^{-2})$

\_\_\_\_\_ 10.  $(3.2 \times 10^2) (2.0 \times 10^{-5}) (3 \times 10^4)$

\_\_\_\_\_ 11.  $(3.80 \times 10^4) \div (1.25 \times 10^{-3})$

\_\_\_\_\_ 12.  $(2.5 \times 10^{-7}) (5.0 \times 10^{-8})$

\_\_\_\_\_  $(9.5 \times 10^{-14}) (6.0 \times 10^8)$

\_\_\_\_\_ 13.  $(389\ 000\ 000) (0.000\ 75)$

\_\_\_\_\_ 14.  $(0.000\ 686\ 8) \div (87\ 000)$

\_\_\_\_\_ 15.  $(3.5 \times 10^{18}) (1.47 \times 10^6) (3.442 \times 10^{-3}) (9.97 \times 10^5)$

\_\_\_\_\_  $(9 \times 10^{31}) (6.634 \times 10^8) (2.7 \times 10^4) (6.02 \times 10^{23})$